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## Polymorphism and function of some oxidizing enzymes of Scots pine (*Pinus sylvestris* L.)\*

### INTRODUCTION

Under investigation were enzymatic proteins from tissues of Scots pine with peroxidase (PO), polyphenoloxidase (PPO) and indole-3-acetic acid oxidase (IAA oxidase) activity. These are enzymes commonly found in plant tissues.

Peroxidases (donor  $H_2O_2$  oxidoreductase E.C. 1.11.1.7.) are nonspecific enzymes with respect to substrates, which in the presence of  $H_2O_2$  can oxidize phenolic compounds, cytochrome c, ascorbic acid, indole, amines and some inorganic ions. They are readily soluble but they occur also in forms bound with cell structures, namely with the cell wall (Ridge and Osborne 1970), ribosomes (Pennon et al. 1970, Darimont and Baxter 1973), mitochondria (Darimont and Baxter 1973), provacuolar membranes (Parish 1975) and various unidentified membranes (van der Mast-1970, Darimont et al. 1977). Activity of PO was also found in chromosomes (Raa 1973).

Polyphenoloxidases are a group of enzymes with so far an undefined function. They oxidize phenolic substrates using molecular oxygen. According to the recommendations of the International Biochemical Union from the year 1964 these enzymes have been divided into two groups: 1) catechol-ortho-diphenol oxidases: oxidoreductase (E.C. 1.10.3.1.) and 2) para-diphenol laccase: oxidoreductase (E.C. 1.10.3.2.). Later a revision of nomenclature was made combining these enzymes into E.C. 1.14.18.1 monophenol: monooxygenase. Mayer and Harel (1979) believe that this revised nomenclature is inappropriate since catechol oxidase and laccase differ substantially in catalytic properties. Catechol oxidase referred to also as phenolase, polyphenoloxidase, catecholase, tyrosinase or crezolase catalyzes two reactions (Mayer and Harel 1979): a) the

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transfer of oxygen from the ortho position to an existing hydroxyl group (crezolase activity), and b) oxidation to ortho-diphenol detaching hydrogen (catecholase activity). Laccase catalyses the reaction of dehydrogenating monophenols, triphenols, ortho-diphenols and as the only oxidase capable of doing it, para-diphenols. Catechol oxidase is easily detectable and thus it was one of the first of plant enzymes to be studied (Bertrand 1895 - 1896). This enzyme has been found in the soluble fraction of plant cells (Mayer 1961, Harel et al. 1964, Sanderson 1964, Coombs et al. 1974), in fractions bound with the cell wall (Palmer 1963, Delon 1974, Stom et al. 1975) and with membranes of organelles (Mayer and Friend 1960, Mayer 1961, Harel et al. 1964, Stafford 1969, Tolbert 1973, Kato et al. 1976).

Indole-3-acetic acid oxidase is an enzyme catalysing the destruction of IAA and it is known to be a ferroprotein. In the literature the majority of data indicates that generally various isoperoxidases are capable of oxidizing IAA though Sequeira and Mineo (1966) and van der Mast (1969) have isolated enzyme preparations capable of oxidizing IAA but demonstrating no PO activity.

The functions of PO, PPO and IAA oxidase have not been strictly determined yet. In vascular plants PO and PPO most probably participate in the production of wood through the oxidation of phenols which then spontaneously condense to lignin, though there is little evidence supporting this function. Freudenberg et al. (1958) have demonstrated the participation of laccase in the oxidative or dehydrogenative polymerisation in vitro of coniferyl alcohol molecules or else of these molecules with those of synapyl alcohol or p-coumaryl alcohol. Since PO in the presence of  $H_2O_2$  can change p-coumaryl alcohol into polymers similar to lignin, as PPO in the presence of oxygen, Higuchi (1957) and Lyr (1957) have considered PO to be lignifying factor in plants. Harkin and Obst (1973) have shown that in the early stages of lignification PO is the only enzyme oxidizing phenols.

Laccase participates in the demethoxyfication of lignin during its degradation (Kirk et al. 1977) and in the case of pathogens it can detoxify fungicidal compounds produced by the host plant (Ward and Stoessl 1973, Rubin and Artsikovskaja 1964).

There is much data in the literature indicating the participation of oxidases in IAA metabolism. Thimann (1934) was the first to propose the theory that the phenol-phenoloxidase system acts inhibively on auxin and Tomaszewski (1959) has found that the same enzymatic system may degrade IAA both in vivo and in vitro forming auxin-phenol complexes (Leopold and Plumer 1961).

The importance of PO in regulation of growth and development of plants has been frequently suggested in the literature. During the development of plants there occur changes in the pattern of electrophoretic

fractions for PO and in the enzymatic activity of specific fractions. There is relatively little information on the changes in isoenzyme patterns during development of forest trees (Durzan 1966, Conkle 1971, Mayberry and Feret 1977). To determine the role of isoperoxidases in the development of plants the isoenzyme patterns have to be compared from extracts of developing tissues obtained from normal plants with those from an undifferentiated callus (Bajaj et al. 1973, Hagiuchi 1977a, 1977b).

The pattern of electrophoretic fractions of oxidases also undergoes changes under the influence of growth substances however the mechanisms by which these compounds act are as yet unexplained. An increase in PO and PPO activity is observed under stress conditions (injury to tissue, action of pathogens, fumigation with industrial pollutants, low humidity, lack of mineral and hormonal substances) and during ageing. It is believed that an increase in activity is associated with a destruction of cellular structures and the liberation of oxidases from inactive forms (Farkas et al. 1964). Possibly also a change of the oxidases occurs from the bound condition with the cell structures into the free form (Volk et al. 1977, 1978, Harel et al. 1966, Kindron et al. 1978).

There are various opinions concerning interrelationships between the activities of PO, PPO, and IAA oxidase: 1. The activity of PO, PPO and IAA oxidase are associated with the same protein molecule (Pilet and Sevhonkian 1969, Janssen 1970, Srivastava and van Huystee 1973, Kieliszewska-Rokicka 1980), 2. The activities of PO and IAA oxidase are strictly connected with each other while PPO is a clearly distinct enzyme (Shinshi and Noguchi 1975), 3. Some plant species do not have PPO (Ray 1960, Bastin 1964, Cambie and Bocks 1966, Dinant and Gaspar 1967).

In a previous investigation (Kieliszewska-Rokicka 1980) the results of electrophoretic studies on PO and PPO in needles and roots of Scots pine have led to the suggestion that pine does not have a specific polyphenoloxidase. In the present study an analysis of the substrate specificity of electrophoretic forms from various organs of pine and from various cell fractions as well as from various chromatographic fractions was aimed at the confirmation or rejection of the view that pine lacks a PPO and the determination of the possible direction of oxidative action in this species.

#### MATERIAL AND METHODS

Plant material. Vegetative organs of several years old Scots pine (*Pinus sylvestris* L.) trees from an experimental forest and of young seedlings growing in a greenhouse at a 16 h photoperiod were investi-

gated. The seeds were sown in perlite acidified with 0.1 N phosphoric acid to pH 7 and then watered with a mineral nutrient solution according to Laiho (1970). For some analyses commercial preparation of purified peroxidase of horse-radish (HRP, lyophilized, Calbiochem.) was used.

Enzyme preparations. 1. The crude extract was prepared on the basis of a method given by Hadačová and Sahulka (1967). The tissue was homogenized in 0.2M phosphate buffer, pH 6, which contained 0.2% ascorbic acid, 0.1% L-cysteine hydrochloride monohydrate and 5% polyclar AT. The homogenate was centrifuged at  $10\,000\times g$ , the supernatant dialysed in 0.01 phosphate buffer, pH 6 and then used as an enzyme preparation. 2. Acetone powder. The tissue was infiltrated under vacuum with a solution containing 0.1M ascorbic acid and 0.1M potassium pyrosulphite and then homogenized in cold 80% acetone and washed with anhydrous acetone to obtain a white powder. From the acetone powder proteins were washed out with 0.1M borate buffer, pH 8 or 0.2M phosphate buffer pH 6. After centrifuging at  $10\,000\times g$  the supernatant was dialysed in 0.01M borate buffer or 0.01M phosphate buffer respectively, and then used as an enzyme preparation.

Extraction of soluble bound enzymes. Pine tissues have been fractionated according to the method given by Birecka and Miller (1974) into: a) buffer soluble proteins, b) proteins soluble in 1% Triton X-100, c) proteins ionically bound with cell membranes, d) proteins covalently bound with cell membranes. All fractions were dialysed in 0.01M phosphate buffer pH 6.

Determination of protein content. The method given by Potty (1969) with albumine of bovine blood serum as standard was used.

Ion-exchange chromatography. The enzyme preparation in 0.1M borate buffer was fractionised in DEAE cellulose column ( $30\times 2$  cm) according to the method of Shinshi and Noguchi (1975).

Enzyme assays. PO activity was determined according to the method of Quoirin et al. (1974) with modifications. The reaction mixture contained 1 ml of 0.2M phosphate buffer, pH 6.1, 1 ml of 0.2M  $H_2O_2$ , 1 ml of 1% quaiacol and a suitable amount of enzyme. The activity was assayed at 430 nm. To determine the activity of IAA oxidase the decomposition of IAA by the enzyme preparations was investigated, using the method given by Shinshi and Noguchi (1975) with Salkowski reagent (Gordon and Weber 1951).

Electrophoresis. Polyacrylamide disc gel electrophoresis in anionic (Davis 1964) and cationic (Reisfeld et al. 1962) systems was carried out. Comparable amounts of protein were applied onto gel columns: 2  $\mu g$  for peroxidase, 200  $\mu g$  for IAA oxidase and polyphenoloxidase. Separation was conducted for 2.5 - 3 hours using 2 mA per one gel column 5 mm in diameter.

Staining of isozymes on polyacrylamide gels. The incubation mixture for PO contained 0.03M phenolic substrates as indicated in figures dissolved in an acetate buffer pH 5 and 0.02M  $H_2O_2$  (modified method of S a h u l k a 1970). Benzidine was used in a saturated solution (S a f o n o v et al. 1969). The reaction was performed for 15 minutes at room temperature. PPO was stained with 0.03M phenolic substrates in 0.2M phosphate buffer, pH 6 (S a h u l k a 1970), or in the same buffer containing 60% ethanol. Some gel columns after 30 minutes of incubation in the buffer solution of substrate were incubated for another 30 minutes in 0.1% solution of p-phenylenediamine or m-phenylenediamine in order to obtain higher sensitivity of reaction (S h e e n and C a l v e r t 1969, v a n L o o n 1971). The oxidations of IAA and L-tryptophan were carried out in IAA and L-tryptophan respectively in the presence of  $MnCl_2$  and 2,4-dichlorophenol with or without  $H_2O_2$  (F r e n k e l 1972). After staining the gel columns were scanned in a Vitatron densitometer.

## RESULTS

### 1. SUBSTRATE SPECIFICITY OF ENZYMATIC PREPARATIONS FROM PINE AND A COMMERCIAL PEROXIDASE PREPARATION FROM THE ROOT OF HORSE-RADDISH

#### 1.1. Pine tissues

On acrylamide in an anionic and cationic systems the enzymatic preparations from various Scots pine tissues were electrophoretically separated. Enzymatic reactions were conducted in order to oxidize various phenolic and indolic substrates in the presence of  $H_2O_2$  and without it. Figure 1 presents the patterns of electrophoretic fractions with an oxidase activity obtained from an enzyme preparation of vegetative buds of pine. The ability to oxidize IAA, L-tryptophan, p-cresol and benzidine was investigated. The oxidation of IAA occurred both in the presence of  $H_2O_2$  and in atmospheric oxygen and the enzymatic fraction catalyzing these reactions has the same Rf value as the fractions oxidizing L-tryptophan, p-cresol and benzidine. L-tryptophan was oxidized exclusively in the presence of  $H_2O_2$ . The product of L-tryptophan oxidation had after a reaction with dimethylaminocinnamaldehyde a pink reaction similarly as the product of IAA oxidation. A similar result has been obtained earlier (K i e l i s z e w s k a - R o k i c k a 1980) when L-tryptophan was oxidized in the presence of  $H_2O_2$  by electrophoretic fractions from poplar tissues.

In a solution of p-cresol without  $H_2O_2$  on the gel two small darkly staining bands formed, however this reaction was not reproduced in the course of studies with other enzyme preparations. On the other hand, in the same solution of p-cresol in the presence of  $H_2O_2$  the electrophoretic fractions have formed a stainless precipitate which suggests the dehydrogenation of a monophenol and its condensation into larger entities.

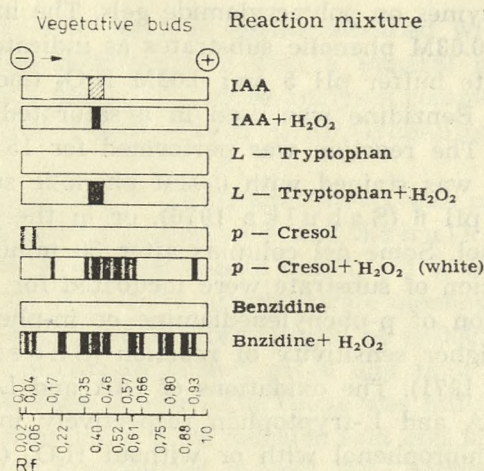


Fig. 1. Electrophoretic separation of an enzyme preparation from Scots pine vegetative buds on acrylamide gels at pH 8.3. The oxidation of IAA and L-tryptophan follows in the presence of  $MnCl_2$  and 2,4-dichlorophenol in 0.2 M phosphate buffer pH 4.5. The staining mixtures for polyphenoloxidase contained 0.03 M p-cresol in 0.2 M phosphate buffer pH 6 and benzidine in the same buffer in a saturated solution. The staining mixtures for peroxidase contained 0.03 M p-cresol and benzidine in 0.2 M acetate buffer pH 5 and 0.02 M  $H_2O_2$ .

In benzidine solution with  $H_2O_2$  added there formed on the gel 12 brownish bands with PO activity. Without  $H_2O_2$  no colour reaction was observed.

In further investigations the ability of the enzyme preparations from pine to oxidize various monophenols and polyphenols was studied in the presence of  $H_2O_2$  and with atmospheric oxygen only. Enzymatic incubation in buffer solutions of phenols without  $H_2O_2$  did not form any colour reactions. Thus for the enzymatic incubation we have used compounds which are known from the literature to couple with the products of dehydrogenation of phenols, namely m- and p-phenylenediamine and ethanol.

Figure 2 presents the electrophoretic patterns for the oxidases of cambium. In the presence of  $H_2O_2$  the studied enzymatic preparation catalysed the dehydrogenation of polyphenols into staining compounds and monophenols into a colourless precipitate. The presence in the incubation mixture of 60% ethanol, 0.1% m-phenylenediamine or 0.1% p-phenylenediamine have resulted in the formation of staining products in spite of the absence of  $H_2O_2$ .

## 1.2. Peroxidase preparation from the root of horse-radish

For comparison the oxidation capacity of commercial peroxidase preparation from root of horse-radish (HRP) was analysed. Figure 3 shows the results of incubation of the HRP fractions separated on acry-

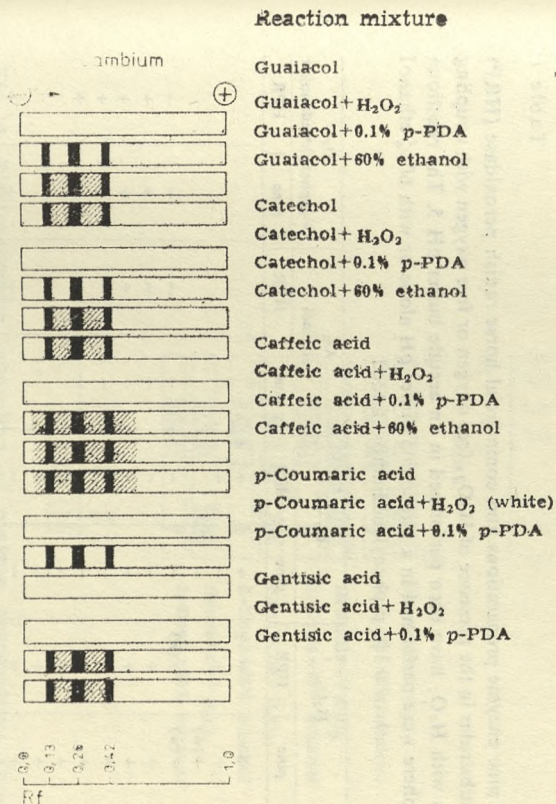


Fig. 2. Electrophoretic separation of an enzyme preparation from Scots pine cambium on acrylamide gels at pH 8.3. Staining mixtures for peroxidase contained phenolic substrates in 2M acetate buffer pH 5 and 0.02 M H<sub>2</sub>O<sub>2</sub>. Polyphenoloxidase activity was detected by incubating the gels in mixtures containing 0.03 M substrates in 0.2 M phosphate buffer pH 6. Some of reactions occur in the presence of 0.1% p-phenylenediamine (p-PDA) or 60% ethanol

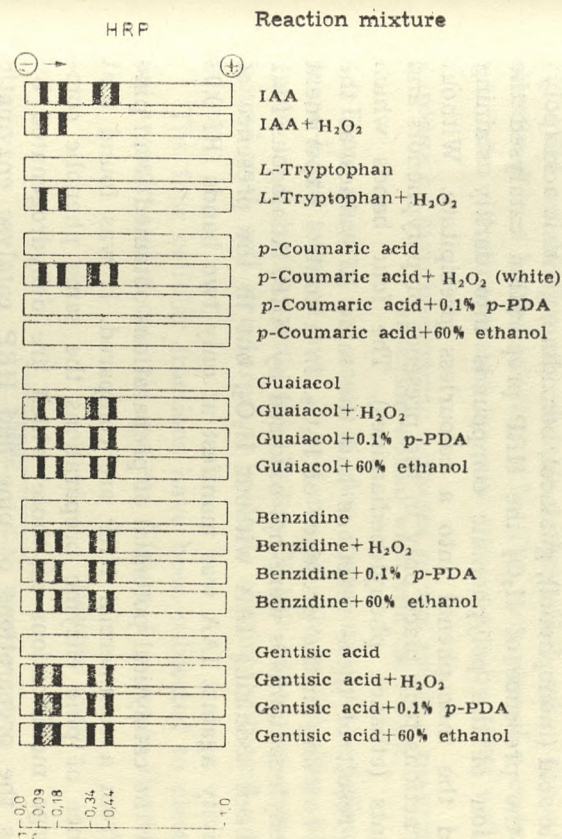


Fig. 3. Electrophoretic separation of horse-radish peroxidase (HRP) on acrylamide gels at pH 8.3. Capabilities of electrophoretic fractions to oxidize IAA and L-tryptopham have been detected as in Fig. 1. The staining mixtures for polyphenoloxidase were used as in Fig. 2

lamide in solutions of the following substrates: IAA, L-tryptophan indoles, p-coumaric acid (monophenol), guaiacol, benzidine, gentisic acid (polyphenols). In the presence of  $H_2O_2$  the HRP preparation catalysed the dehydrogenation of the polyphenolic compounds into darkly staining products, and the monophenol into a colourless precipitate. Without  $H_2O_2$  these reactions took place only in the presence of polyphenols and coupling agents (ethanol, phenylenediamines). The four bands which formed as a result of these reactions had the same Rf values as the bands of PO reacting in the presence of  $H_2O_2$ . In a control experiment m- and p-phenylenediamines were not oxidized by HRP. Analogous four bands have been oxidizing IAA without  $H_2O_2$ , but in the presence of  $H_2O_2$  the activity against IAA was manifest at only two bands (Rf 0.08 and 0.18).

In Table 1 the catalytical potential of preparations obtained from pine tissues and from a commercial HRP are compared. It was found that in the presence of pine enzyme preparations the same phenolic compounds undergo modifications as those which are dehydrogenated by HRP. Besides the preparations of pine and HRP catalyse enzymatic reactions only in some, already identical conditions. The IAA oxidation by peroxidase in the presence of atmospheric oxygen has been known for some time (K e n t e n 1955, R a y and T h i m a n n 1956). The present results indicate the possibility of oxidation of other substrates also by PO of pine and of horse radish without  $H_2O_2$  when in the reaction mixture ethanol or phenylenediamine are present. Pine and HRP prepara-

Table 1

Oxidative activity of pine enzyme preparations and commercial horse radish peroxidase (HRP) on various phenolic substrates in the presence of  $H_2O_2$ , free oxygen or free oxygen with coupling agents. The reactions with  $H_2O_2$  have been performed in an acetate buffer pH 5. The reactions with  $O_2$  from atmosphere were performed in a phosphate buffer pH alone or with 60% ethanol or 0.1% p-phenylenediamine added

Substrates	$H_2O_2$		$O_2$		$O_2$ + ethanol		$O_2$ + p-phenylenediamine	
	pine	HRP	pine	HRP	pine	HRP	pine	HRP
ndole-3-acetic acid	+	+	+	+				
L-tryptophan	+	+	-	-				
p-cresol	+(w)		-	-				
p-coumaric acid	+(w)	+(w)	-	-	-	-	-	-
Catechol	+	+	-	-	+	+	+	+
Guaiacol	+	+	-	-	+	+	+	+
Caffeic acid	+	+	-	-	+	+	+	+
L-dopa	+	+	-	-	+	+	+	+
Chlorogenic acid	+	+	-	-				
Hydroquinone	+	+	-	-				
Gentisic acid	+	+	-	-		+	+	+
Benzidine	+	+	-	-	+	+	+	+
Pyrogallol	+	+						

+ positive reaction, - no reaction w white precipitate



tions did not oxidize any phenols without  $H_2O_2$  when the substrates were dissolved in a buffer solution only. The monophenol p-coumaric acid has not been dehydrogenated by pine tissues nor by HRP without  $H_2O_2$ .

## 2. FORMS OF ELECTROPHORETIC OXIDASES IN THE SOLUBLE FRACTION AND IN THE FRACTION BOUND WITH THE CELL WALL

From needles and roots of pine consecutively the following fractions were being washed out: a) proteins soluble in a buffer, b) proteins soluble in a buffer with 1% Triton X-100 added, c) proteins ionically bound with cell structures, d) proteins covalently bound with cell structures.

The four cellular fractions have been subjected to electrophoresis in an anionic and a cationic system. Figure 4 presents the electrophoretic

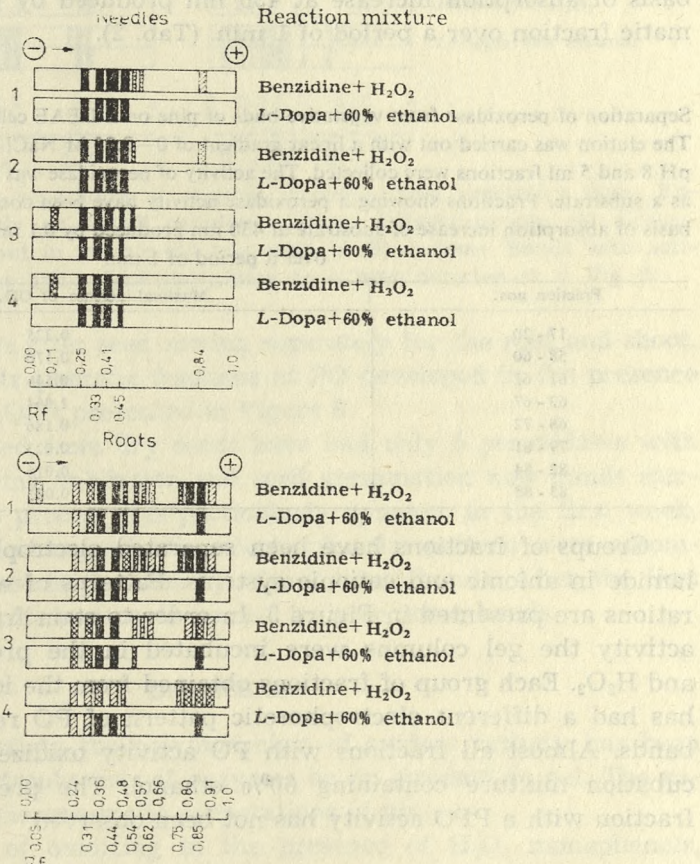


Fig. 4. Electrophoretic separation of enzyme preparations from needles (A) and roots (B) on acrylamide gels at pH 8.3. Pine tissues have been fractionated into: 1 — buffer soluble fraction, 2 — fraction soluble in 1% Triton X-100, 3 — ionically bound fraction, 4 — covalently bound fraction. The staining mixtures for peroxidase and polyphenoloxidase were used as in Fig. 2

bands of PO in individual cell fractions obtained as a result of incubation of gels in a solution of benzidine and  $H_2O_2$  and of bands obtained through incubation of columns in a solution of an ortho-diphenol L-dopa in the presence of ethanol. The majority of PO bands have been oxidizing also L-dopa, but in none of the cellular fractions was the presence of a specific PPO demonstrated.

### 3. ELECTROPHORETIC FORMS WITH PO AND PPO ACTIVITY IN ENZYMATIC PREPARATIONS FRACTIONATED ON AN ANION EXCHANGE COLUMN

The protein preparation from vegetative buds of pine has been separated in a column of ion-exchange cellulose Cellex-D in NaCl gradient. Fractions 5 ml in volume were collected and in the individual fractions the activity of PO was determined using guaiacol as a substrate. Fractions showing a PO activity have been combined into groups on the basis of absorption increase at 430 nm produced by 0.1 ml of the enzymatic fraction over a period of 1 min. (Tab. 2).

Table 2

Separation of peroxidase from vegetative buds of pine on a DEAE cellulose column ( $2 \times 30$  cm). The elution was carried out with a linear gradient of 0–0.25 M NaCl in 0.01 M Tris-HCl buffer pH 8 and 5 ml fractions were collected. The activity of peroxidase was determined using guaiacol as a substrate. Fractions showing a peroxidase activity have been combined into groups on the basis of absorption increase of substrate at 430 nm produced by 0.1 ml of the enzymatic fraction over a period of 1 min

Fraction nos.	Maximal increase of $OD_{430}$ /l/ 0.1 ml
17 - 20	0.325
58 - 60	0.170
61 - 62	0.546
63 - 67	1.864
68 - 72	0.186
79 - 81	0.023
82 - 84	0.016
85 - 88	0.023

Groups of fractions have been separated electrophoretically on acrylamide in anionic and cationic systems. Patterns of electrophoretic separations are presented in Figure 5. In order to stain fractions having a PO activity the gel columns were incubated in the presence of benzidine and  $H_2O_2$ . Each group of fractions obtained from the ion-exchange column has had a different electrophoretic pattern of PO represented by 2 - 12 bands. Almost all fractions with PO activity oxidized L-dopa in an incubation mixture containing 60% ethanol. The presence of a specific fraction with a PPO activity has not been observed.

### 4. ISOPEROXIDASES IN THE EARLY STAGES OF PINE TREE DEVELOPMENT

Enzymatic preparations from germinating embryos, seeds and developing seedlings have been separated on acrylamide in an anionic system. Initially the enzymatic preparations have been made from the whole

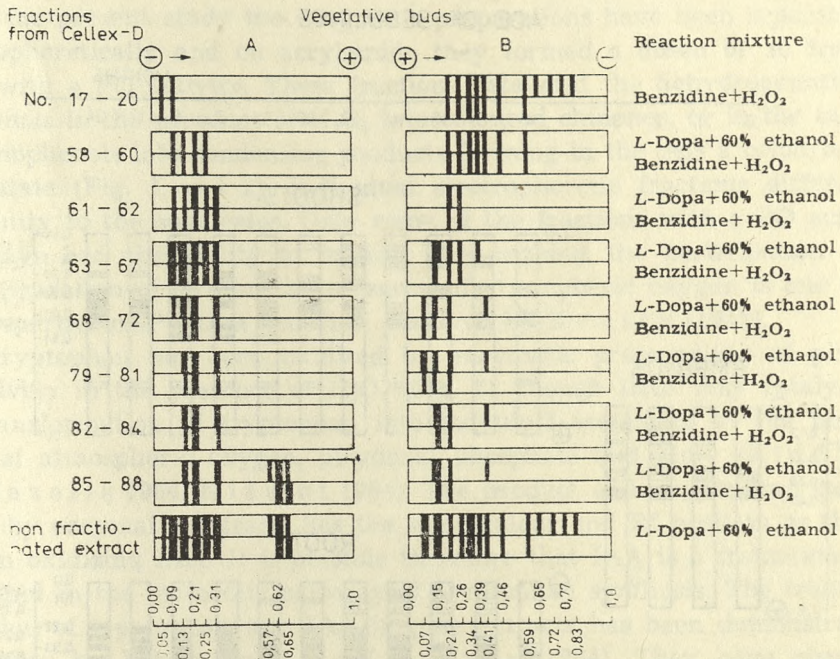


Fig. 5. Electrophoretic separation of enzyme fractions from vegetative buds. Extracts eluted previously on DEAE cellulose column. Polyacrylamide gel electrophoresis was carried out in anionic (A) and cationic (B) systems. Bands with activities of peroxidase and polyphenoloxidase have been detected as in Fig. 2

embryos, and 4 days after seed sowing separately for the root and shoot. The pattern of electrophoretic fractions of PO developed in the presence of benzidine and H<sub>2</sub>O<sub>2</sub> is presented in Figure 6.

Embryos dissected from dry seeds have had only 5 peroxidases with a low activity. During imbibition and seed germination new bands started to appear. This process was particularly dynamic in the first week, and the qualitative changes in the electrophoretic pattern were accompanied by an increase in the activity of individual bands. After the first week of growth shoots and roots tended to lose some bands.

#### DISCUSSION

In the present study the polymorphism of oxidase activity has been analysed using electrophoresis of enzymes on an acrylamide gel. The results indicate that the enzymatic preparations of pine:

1. Are capable of oxidizing in the presence of H<sub>2</sub>O<sub>2</sub> monophenols, o-polyphenols, p-polyphenols, IAA and L-tryptophan (peroxidase activity),
2. Are capable of oxidizing IAA in the presence of atmospheric oxygen (IAA oxidase activity),



In the present study the enzymatic preparations have been separated electrophoretically and on acrylamide they formed a dozen or so fractions with a PO activity. These fractions catalysed the dehydrogenation of phenols in the presence of  $H_2O_2$  into coloured chinones, or in the case of monophenols into condensing products forming in the gels a colourless precipitate (Fig. 1 and 2). Individual electrophoretic fractions differed in affinity to the substrates. Only some of the fractions with a PO activity have had the ability to oxidize IAA without the participation of  $H_2O_2$ . Oxidation of IAA in the presence of atmospheric oxygen is one of the properties of PO that was first observed by K e n t e n (1955).

L-tryptophan has been oxidized by enzymatic preparations of pine exclusively in the presence of  $H_2O_2$  (Fig. 1), though HRP may catalyse the transformation of tryptophan into indolilo-3-acetamide in the presence of atmospheric oxygen, pirydoxal phosphate and  $Mn^{2+}$  (Riddle and M a z e l i s 1964, K l ä m b t 1964). The product of L-tryptophan oxidation by enzymatic extracts has the same colour and Rf position as the fraction oxidizing IAA. It is possible therefore that IAA is a transitional compound in the metabolic pathway of tryptophan synthesis. The transformation of tryptophan into IAA in pine extracts has been demonstrated earlier by W h i t e m o r e and Z a h n e r (1964). They have given tryptophan to extracts of cambium and phloem of pine and they have found the formation of IAA. In fresh extracts IAA formed rapidly, but it also formed, though much more slowly in boiled extracts. Since PO can maintain its activity even after boiling it is possible that it was this enzyme that catalyzed the transformation of tryptophan to IAA. Decarboxylation of IAA and L-tryptophan has been observed when labelled substrates were introduced into seeds and pine explants (H e j n o w i c z and T o m a s z e w s k i 1969, T o m a s z e w s k i 1970a, 1970b).

One of the possible physioecological functions of PO associated with the cell wall could be the degradation of exogenous auxin. When penetrating into cell walls PO could be disturbing the processes controlled by endogenous auxin, as is the case in the mycorrhizal associations of pine (T o m a s z e w s k i and W o j c i e c h o w s k a 1974).

In the presence of atmospheric oxygen electrophoretic fractions of pine tissues have not been transforming any phenols into colour products, even though they have produced intensive colour reactions when the incubation took place in the presence of compounds coupling with products of the oxidation of phenols such as m-phenylenediamine, p-phenylenediamine or when the substrate was dissolved in a buffer-ethanol mixture (Fig. 2). S h e e n and C a l v e r t (1969) and v a n L o o n (1971) when using the factors associating themselves with the products of phenol oxidation have obtained an increase in the intensity of colour reactions catalyzed by PPO.

In the present study at no stage did it prove possible to physically

separate the PPO activity from that of PO. Enzymatic proteins with a PO activity have been found in soluble fractions and in those bound ionically or covalently with cell structures. Treating cell structures with a Triton X-100 solution after having washed away proteins soluble in a buffer has demonstrated the presence of a very active PO (Fig. 4). Similar effects have been obtained by Walker and Hulme (1966) and by Harel and Mayer (1971). According to some authors the strength of the bond between oxidases and membranes depends on the type of tissue and the state of plant development. In none of the four cell fractions studied in the present investigation was the existence of a specific PPO activity demonstrated. Also the double fractionation of pine enzyme preparations first in anion-exchange cellulose and then electrophoretically on acrylamide (Fig. 5) did not disclose any protein forms with a specific PPO activity.

On the basis of results obtained in the present study one can suspect that PPO activity in pine is very low or that the studied enzyme preparations did not contain PPO and the reactions of dehydrogenating phenolic substrates can in some conditions take place irrespective of this enzyme.

Specific electrophoretic fractions with an exclusively PPO activity have been found in other plants for example in tobacco (Shinshi and Noguchi 1975) or in poplar bark (Kieliszewska-Rokicka 1980). On the other hand, Srivastava and van Huystee (1973) have shown electrophoretically and chromatographically that tissue cultures of peanuts contain true PPO which however appears in the same fractions as PO and IAA oxidase.

Pine tissues have shown the same catalytic abilities as the commercial preparation of a purified PO of horse radish root (HRP). Figure 3 and Table 1 indicate that in some conditions HRP catalyzes oxidation reactions of phenolic compounds without the participation of  $H_2O_2$ , though it does not contain PPO. One can suspect therefore that also the pine tissue preparations studied here behave in the same way as HRP, that is they can in certain special conditions catalyze the oxidation with free oxygen.

In spite of the fact that PPO is common in nature, lack of this enzyme has been found in *Osmunda* sp. (Ray 1960), *Impatiens* (Bastin 1964) and in *Phaseolus* sp. (Dinant and Gaspar 1967). In all these plants PO and IAA oxidase were active. The lack of PPO has been also found by Cambie and Bock (1966) in 24 out of 44 studied *Gymnospermae* species. These authors have not studied the genus *Pinus*, but they have found the absence of PPO in *Cedrus libani* which belongs to the *Pinaceae*.

Shain and Mackay (1973) have identified the activity of o-diphenoloxidase and p-diphenoloxidase in the heart-wood of *Pinus radiata*.

They have measured manometrically the absorption of oxygen by fresh and boiled heart-wood fragments incubated with phenolic substrates. Their results indicated that the activity of phenoloxidase originates rather from the pine tissues than from microorganisms. The occurrence in the heart-wood of *Pinus lambertiana* of brown stains suggesting a PPO activity has been observed by Stutz (1959). However Anderson et al. (1963) and Higuchi et al. (1967) have suggested that the oxidation of phenolic substrates in the heart-wood is a nonenzymatic process (compare with Tomaszewski 1967).

It has long been known that besides oxidation in the presence of  $H_2O_2$  in the reaction mixture PO is capable of oxidizing various compounds in the presence of air. So far such oxidation reactions have been described for dihydroxyfumaric acid (Swedin and Theorell 1940, Chance 1952), IAA (Kenten 1955, Ray and Thimann 1956), triose reductone (Yamazaki et al. 1965), and naphthohydrochinon (Klapper and Hackett 1963a, 1963b). Besides PO is capable of hydroxylation of aromatic molecules in the presence of atmospheric oxygen and dihydroxyfumaric acid (DHF) (Mason et al. 1957, Buhler and Mason 1961). In this reaction PO may be replaced by some other heme system. The hydroxylation reaction can be also obtained in the presence of atmospheric oxygen without enzymatic participation. Such a reaction has been obtained by Undenfriend et al. (1954) in the reaction mixture: inorganic iron ( $Fe^{2+}$  or  $Fe^{3+}$ ) + ascorbic acid + EDTA. In the presence of atmospheric oxygen PO may also catalyze the formation of ethylene from methional (Yang 1967). Here similarly as in the case of IAA oxidation monophenols stimulate the reaction and polyphenols inhibit it (Yang 1969), and it is also known to have antiseptic activity (Klebanoff 1967, 1968, Lehrer 1969).

In view of these oxidative abilities one can suspect that the slow dehydrogenation of phenolic compounds in the presence of atmospheric oxygen is an unusual property of PO isolated from pine tissues.

There exists an opinion that one of the functions of PPO is to synthesize o-diphenols. Vaughan et al. (1969) have found that catecholase can hydroxylate naturally occurring phenols. However the hydroxylation of phenols as was mentioned earlier may be catalyzed by PO or have a nonenzymatic course. Stafford (1969) has found an active synthesis of diphenols from endogenous precursors and in the presence exogenous phenylenediamine or tyrosine, the PPO activity simultaneously declining.

Since PO may cause a change in the concentration of IAA, a growth hormone, it is suggested in the literature that there is a relation between the activity of PO and the growth of plants (Ockerse et al. 1966). Histochemical studies have allowed the conclusion that PO is strictly connected with the zone of actively dividing cells or with cells that are

just prior to division (Vanden Born 1963, Van Fleet 1959, De Jong 1967).

In the present study, in the first week of embryo development after seed imbibition, when the developmental processes are very intense, it was observed that there is a dynamic increase in the number and activity of electrophoretic fractions of PO (Fig. 6). Similar increases have been observed in the early development of *Pinus attenuata* (Conkle 1971) and *Pinus banksiana* (Ramaiah et al. 1971), though Gordon (1971) has shown that a high PO activity accompanied slow growth of *Populus deltoides* (Bartr.).

#### SUMMARY

Under investigation were electrophoretically fractionated oxidizing enzymes of Scots pine. Substrate specificity for the enzyme complex (peroxidase, polyphenoloxidase, and indole-3-acetic acid oxidase) was studied and compared to that of a purified peroxidase of horse radish. The studies have shown that plant tissues contain proteins with a peroxidase activity capable of dehydrogenation o-polyphenols and p-polyphenols to coloured products in the presence of  $H_2O_2$ . Monophenols undergo dehydrogenation and then condense forming a colourless precipitate. With the help of colour reactions it was found that peroxidase of pine tissues has the capacity to oxidize indole-3-acetic acid to 3-methylenoxyindole both in the presence of  $H_2O_2$  and in the presence of atmospheric oxygen. L-tryptophan is oxidized by pine peroxidase exclusively in the presence of  $H_2O_2$ . One can suspect that indole-3-acetic acid is a transitional product of this reaction. In the absence of  $H_2O_2$  purified protein preparations from pine tissues dehydrogenate phenolic compounds only in the presence of coupling factors (m- and p-phenylenediamines, ethanol).

A commercial purified preparation of horse radish root peroxidase (HRP) has the same catalytical properties as the preparations from pine tissues. In preparations of pine tissues the activity of polyphenoloxidase cannot be physically separated from that of peroxidase. Enzymatic proteins having an oxidase activity occur in soluble fractions and in fractions that are ionically or covalently bound with cell structures. Specific fractions have been demonstrated electrophoretically which have peroxidase activity, peroxidase and IAA oxidase activity and peroxidase and polyphenoloxidase activity.

The conclusion was reached that pine tissues do not have specific polyphenoloxidases and that in special conditions proteins containing a specific peroxidase can, similarly as peroxidase of horse radish root, catalyze the oxidation of substrates with free oxygen.



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## Wielopostaciowość i funkcja niektórych enzymów utleniających sosny zwyczajnej (*Pinus sylvestris* L.)

### Streszczenie

Przedmiotem pracy było elektroforetyczne rozfrakcjonowanie enzymów utleniających sosny zwyczajnej. Badano specyficzność substratową kompleksu enzymów (peroksydaza, polifenoloksydaza i oksydaza kwasu indolilo-3-octowego) i porównywano ją ze specyficznością oczyszczonej peroksydazy korzenia chrzanu. Badania wykazały, że tkanki sosny zawierają białka o aktywności peroksydazy zdolne do odwodorowania w obecności  $H_2O_2$  orto-polifenoli i para-polifenoli do barwnych produktów. Monofenole ulegają odwodorowaniu i następnie kondensują tworząc bezbarwny osad. Przy pomocy reakcji barwnych stwierdzono, że peroksydaza tkanek sosny ma zdolność utleniania kwasu indolilo-3-octowego do 3-metylenooksyindolu zarówno w obecności  $H_2O_2$  jak i tlenu z powietrza. L-tryptofan jest utleniany przez peroksydazę sosny wyłącznie w obecności  $H_2O_2$ . Wydaje się, że kwas indolilo-3-octowy jest pośrednim produktem tej reakcji. Wobec braku  $H_2O_2$  oczyszczone preparaty białkowe z tkanek sosny odwodorowują związki fenolowe jedynie w obecności czynników sprzęgających (meta i para-fenylendwuamina, etanol). Handlowy preparat oczyszczonej peroksydazy korzenia chrzanu (HRP) wykazuje takie same zdolności katalityczne jak preparaty tkanek sosny.

W preparatach tkanek sosny nie można fizycznie oddzielić aktywności polifenoloksydazy od peroksydazy. Białka enzymatyczne o aktywności oksydaz występują we frakcjach rozpuszczalnych oraz we frakcjach związanych jonowo lub kowalennie ze strukturami komórkowymi. Elektroforetycznie ujawniono specyficzne frakcje peroksydazy, frakcje o aktywności peroksydazy i oksydazy IAA oraz frakcje o aktywności peroksydazy i polifenoloksydazy.

Wprowadzono wniosek, że tkanki sosny nie posiadają specyficznej polifenoloksydazy, a w specjalnych warunkach preparaty białkowe zawierające specyficzną peroksydazę mogą, podobnie jak oczyszczona peroksydaza korzenia chrzanu, katalizować utlenianie substratów tlenem.

## Полиморфизм и функция некоторых окислительных ферментов у сосны обыкновенной (*Pinus sylvestris* L.)

### Резюме

Предметом исследований было фракционирование методами электрофореза окислительных ферментов сосны обыкновенной. Исследовали субстратную специфичность комплекса ферментов (пероксидаза, полифенолоксидаза, ИУК-оксидаза) и сравнивали ее со специфичностью очищенной пероксидазы корня хрена. Выяснено, что ткани сосны содержат белки с пероксидазной активностью, способные дегидрировать в присутствии  $H_2O_2$ , о-полифенолов и р-полифенолов до окрашенных продуктов. Монофенолы дегидрируют а затем конденсируются образуя бесцветный осадок. С помощью цветных реакций обнаружено, что пероксидаза тканей сосны способна окислять ИУК до 3-метиленооксинидола так в присутствии  $H_2O_2$  как и атмосферного кислорода. L-триптофан окисляется пероксидазой сосны исключительно в присутствии  $H_2O_2$ . Можно судить, что ИУК является промежуточным продуктом в этой реакции. При отсутствии  $H_2O_2$  очищенные белковые препараты тканей сосны дегидрируют фенольные соединения исключительно в присутствии м- и р-фенилдиамина и этанола.

Коммерческий препарат очищенной пероксидазы корней хрена характеризуется

сходными с препаратами с тканей сосны каталитическими способностями. В препаратах с тканей сосны практически не можно физически отделить активности полифенолоксидазы и пероксидазы. Энзиматические белки с оксидазной активностью встречаются в растворимых фракциях и фракциях с ионными и ковалентными связями с клеточными структурами. Методами электрофореза выявлены специфические фракции пероксидазы, а также фракции с пероксидазной и ИУК-оксидазной активностью и фракции с активностью пероксидазы и полифенолоксидазы.

Предполагается, что ткани сосны характеризуются отсутствием специфической полифенолоксидазы, а в специальных условиях белковые препараты содержащие специфическую пероксидазу могут также как и очищенная пероксидаза корней хрена, катализировать окисление субстратов кислородом.